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# Structural and Optical Properties of SnS<sub>2</sub>:Cu Thin films prepared by chemical Spbay Pyrolysis

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# Abqtract:

Thin filis have been prepared from the tin disulphide ( $SnS_2$ ), the pure and the doped with copper ( $SnS_2$ :Cu) with a percentages (1,2,3,4)% by using ahemical spray pyrolysis technique on substrate of glass heated up to(603K)and sith thicknesses (0.7±0.02)µm ,after that the films were treated thermally with a low pressure ( $10^{-3}$ mb) and at a temperature of (473K) for one hour.

The influence of both doping with copper and the thermal treatment on some of the physical characteristics of the prepared films(structural and optical) was studied.

The X-ray analysis showed that the prepared films were polycrystalline Hexagonal type.

The optical study that included the absorptance and transmitance spectra in the weavelength range (300-900)nm demonstrated that the value of absorption coefficient ( $\alpha$ ) was greater than ( $10^4$  cm<sup>-1</sup>) for the pure and doped films and that the electronic transitions at the fundamental absorption edge were of the indirect kind whether allowed or forbidden and the value of the optical energy gap in the case of the indirect transition, the allowed decreased from (2 eV) to (1.8,1.7,1.5,1.2)eV at the doping percentages (1,2,3,4)% respectively, also it was found that the value of energy gap for the pure and doped films increased after annealing.

Tthe absorption and transmission spectra were used to find the optical constant including refractive index(n), extinction coefficient (k), imaginary and real part of dielectric constant ( $\epsilon_{1\,\&}\epsilon_{2}$ ), and it was found that all the optical constant was affected by changing the doping percentages; in addition to being affected after treating the films thermally

## Key word: Optical, Structural Properties, SnS2, SnS2:Cu

## **Introduction:**

The electronic and optical properties of semiconductor materials are tunable varying their shapes and sizes[1,2,3].

The optical properties of semiconductors are often subdivided into those that are electronic and those that are lattice in nature. The electronic properties concern processes

involving the electronic states of the semiconductor, while the lattice properties involve vibrations of the lattice (absorption and creation of phonons). Lattice properties are of considerable interest, but it is the electronic properties which receive the most attention in semiconductors

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because of the technological importance of their practical applications [4].

High absorption region observed for most semiconductors at α≥10<sup>4</sup>cm<sup>-1</sup> · the absorption is due to the transitions between extended states in both bands. The imperial formula that governs this transition have been found by Tauc [5].

Optical properties of SnS<sub>2</sub> thin films were studied by Thangaraju et.al.[6], Ji et.al. [7], Yanng et.al.[8], and Sanchez et.al.[9].

The aim of this research is a preparation of (SnS<sub>2</sub>:Cu) with a percentages (1,2,3,4)% and studying its optical properties. Indeed, the main task was studying the effect of the heat treatment and composition on optical properties of pure SnS<sub>2</sub> and *SnS<sub>2</sub>:Cu* thin films which were prepared by using the chemical spray pyrolysis technique

# Material and methods:-

 $\rm SnS_2$  layers were prepared by spray pyrolysis of aqueous solution of  $\rm SnCl_4.H_2O$  and thiourea. The molar concentration of the solution should be equal to 0.3 mole/liter. In order to prepare the solution of 0.1 molar few grams ,[(2.62935 gm)  $\rm SnCl_4.H_2O$  ,and(0.57093gm)CS(NH2)2]concentrat ions from these two material are weight needed from each of them, melted in 25 liter of distilled water, according to the following equation:

$$M = (W_t / M_{wt}) . (1000/V)$$
 .....(1)

Wt:Weight of the material (gm),V:Volum

# (L),M:Moleculaconcentration (M/L),Mwt:Molecular weight (gm/M)

This composition was optimum to give higher optical transparency. The obtained solution is immediately sprayed with the help of double nozzle sprayer onto heated substrate of glass plates, the upper container of the nozzle has 4 cm diameter and was connected to capillary of 0.127 mm through the stopcock. The capillary was surrounded by a bubble tube through which the compressed air was blown at 2Kg cm<sup>-2</sup>. The sprayer set up has been described elsewhere. The substrate were heated to a temperature of about 603°C for 20 min before spraying in small amounts to avoid excessive cooling of hot substrate during spraying. To obtain films of uniform thickness the distance between sprayer and substrate was kept 30 cm and spray rate was 10 cm<sup>3</sup> min<sup>-1</sup>. Reproducible films were obtained from successive runs. The chemical reaction can be described as:

$$NH_2SCNH_2 + 4H_2O \longrightarrow 2H_2S$$
  
+  $4NH_3 + 2SO_2$   
 $SnCl6 + 2H_2S + 4H_2O \longrightarrow SnS_2$   
+  $H_3O^- + 6Cl^- \dots (2)$ 

To prepared SnS<sub>2</sub>:doped with copper Cu, we added CuCl(1,2,3,4wt%) to SnCl<sub>4</sub>H<sub>2</sub>O and CS(NH<sub>2</sub>)<sub>2</sub> solution. according to the following equation:

$$CuCl_2 \rightarrow Cu + Cl_2\uparrow$$
....(3)

 $SnS_2$  and  $SnS_2$ :Cu films were prepared on glass substra

The thickness of prepared samples were in the range of  $(0.7\pm0.2\mu\text{m})$  which determine using Weight Method and optical Interference method.

the films were clear, transparent and golden-yellow in colour. The coating was found to be stable and has a very good adhesive properties. Optical microscopic studies shows that samples prepared by spray pyrolysis method exhibit very smooth surfaces free from pinholes. To confirm the films purity and structure X-ray diffraction pattern has been taken using  $CuK\alpha$  radiation from a Philips pM

1050/70 diffractometer. The values of interplaner spacing and intensity ration is exactly the same as that of the ASTM data of  $SnS_2$ . the golden-yellow colour of the films becomes grey-black when the samples are heated to  $403\,C$ , mostly probably due to the formation of SnS or  $Sn_2S_3$  phase.

We have carried out detailed analysis of optical data in order to investigate the possible optically allowed transitions at room temperature.

The transmission and absorption spectra were obtained over the range (200-1200)nm by UV-VISIBLE recording spectrometer (Shimadzu modelUV-160).

The optical measurements were repeated on fifteen samples of different thickness and the repetition of results was observed. The reproducibility of data in the films prepared sequentially from the same composition ratio indicates the constancy of composition of the films.

## **Results analysis and discussion:**

1.X-Ray Diffraction: Fig (1) show the XRD patterns obtained for as deposited and heat treated pure SnS2 and doped with (4wt%). According to (ASTM) cards, the structure of the films showed a polycrystalline hexagonal structure. The analysis is demonstrated the reflection surfaces [(001),(002),(003)] and the films have crystallized with a strong peak at (001) directions, while in fig.(1-c,d) the strong peak at(001) directions, this means that this plane is suitable for crystal growth. This results are similar to those obtained by Amalraj et.al.[10], Khelia et.al.[1] and Sanchez et.al.[9] but it disagreement with Kawano et.al.[11]

# 2- The Transmission And Absorption Spectrum

The experimental transmission spectrum for as-deposited and heat treated pure SnS<sub>2</sub> films and doped with

Cu (1,2,3,4wt%) films, respectively, in the wavelenght range (300 – 900)nm at room temperature are shown in Figs.(2).

All spectra show good transparency (T > 85%) and reveal very pronounced interference effects for photon energies below the fundamental absorption edge. They also display a clear explicit absorption edge interrelated to the optical band gap. It is clear from these figures that spectral characterization are affected by heat treatment. This fact is due to the increase in the grain size observed for heat-treated samples[11].

3-Absorption Coefficients: The coefficient absorption α were determined from the region of high absorption i.e. at the fundamental absorption edge of the films using the equation ( $\alpha = 2.303 \text{ (x/d) } [12]$ where d: is the thickness of the film,(x) the absorbance represents the logarithm of the reciprocal (T)).Fig.(3a and b) shows the dependence of the absorption coefficient on the Cu concentration as a function of photon energy for as-deposited and annealed pure SnS<sub>2</sub> films and doped with Cu (1,2,3,4wt%) films .From this figure we can observe that the optical absorption coefficient for all films were fairly high values reached above  $10^4 \text{cm}^{-1}$ , and found that the  $\alpha$  decreases with heat treatment for all samples as shown in (6b), and this is due to the increasing the value of optical energy gap after annealing process. This result agrees with the result of Thangargju et.al.[10].

4- The Optical Energy Gap Measurements: The optical energy gap values  $(E_g^{opt})$  for SnS<sub>2</sub> and SnS<sub>2</sub>:Cu films have been determined by using Tauc <sup>[5]</sup> equation  $(\alpha(v)hv = B(hv - E_g^{opt}))$  where v is the frequency of the incident radiation, B is constant, which depends on density of state (DOS) of

conduction and valence bands,  $E_g^{opt}$  is the optical energy gap <sup>[13]</sup>, r is a constant depend on the nature of the transition.

) which is used to find the type of the optical transition by plotting the relations  $(\alpha h v)^2$ ,  $(\alpha h v)^{2/3}$ ,  $(\alpha h v)^{1/2}$  and  $(\alpha h v)^{1/3}$  versus photon energy (hv) and select the optimum linear part. It is found that the first relation yields linear dependence, which describes the allowed indirect transition.  $E_g^{opt}$  is then determined by the extrapolation of the portion at  $(\alpha h v = 0)$  as shown in Figs.(4a and b).

The optical energy gap  $SnS_2$  was 2eV and these results agree with the results of the energy gap versus concentration studied by Amalraj et.al.[10] and with George et.al.<sup>[14]</sup>.

The optical energy gap was found to increases after heat treatment, this increasing is attributed to existence of a tails below the fundamental absorption edge (near the bands) which observed in the absorption coefficient Fig.(5a), which is a characteristic of most of the optical data on polycrystalline semiconductors which is considered to be determined mainly to the structural disorder existing at the grain boundaries[16], these tails decreased after heat treatment due to the increasing in crystallite size and the grain boundaries became more ordered than in as-deposited films.

# **5- Optical Constants**

**5-1 Refractive Index:** Fig.(6) shows the variation of refractive index with photon energy as-deposited and annealed SnS<sub>2</sub> and SnS<sub>2</sub>:Cu films have been determined by using the equation:

$$n = \left(\frac{4R}{(R-1)^2} - k^2\right)^{1/2} - \frac{(R+1)}{(R-1)}$$

where R is the reflectance[8] k is the extinction coefficient. It is found from this figure that the refractive index

decreases with the increasing of wavelength of the incident photon.

We can observed the appearance of a peak occur in dispersion of the refractive index of the films, this was attributed to the rapid change in the optical absorption coefficient in the vicinity of the absorption edge[14].

The refractive index increases after the heat treatment probably due to the increase of the compactness of the films after the heat treatment simultaneously with the increase of the crystallite size.

#### 5-2 Extinction Coefficient

 $\begin{array}{ccc} \text{The} & \text{extinction} \\ \text{coefficient}^{[15]}, & \text{which} & \text{calculated} & \text{from} \\ \text{the relation:} & k = \frac{\alpha \lambda}{4\pi} \\ \end{array}$ 

The relation between the extinction coefficient and wavelength for SnS<sub>2</sub> and SnS<sub>2</sub>.Cu films deposited at 603K and heat treated at 473K with different values of x is shown in Fig.(6). From this figure it is found that the extinction coefficient (k) takes the similar behavior of the corresponding absorption coefficient.

One can deduce from this figure that the extinction coefficient increased with decreasing the photon energy up to  $\approx 3.647 \mathrm{eV}$  due to the high values of the absorption coefficient at this range of wavelength, after that the extinction coefficient increased very slightly which is associated with the increasing of the transmittance in this region and the large decreasing of the absorption coefficient at this wavelength.

The extinction coefficient is affected by heat treatment, which is decreased after annealing process. This behavior of the extinction coefficients values similar to that of the absorption coefficients for the same reasons as mentioned before.

### 5-3 Dielectric Constant

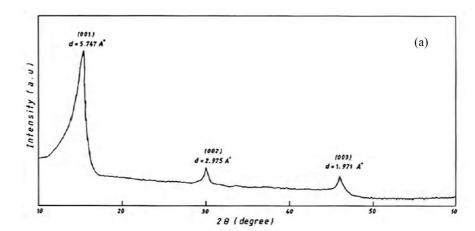
Figs.(7) and (8) show the variation of the real  $(\varepsilon_1)$  which calculated from the relation  $\epsilon_1 = n^2 - k^2$  and imaginary ( $\epsilon_2$ , which calculated from the relation  $\varepsilon_2 = 2nk$ [17] ) parts of the dielectric constant with photon energy as-deposited and andSnS2:Cu annealedSnS<sub>2</sub> films. The behavior of  $\varepsilon_1$  and  $\varepsilon_2$  is the same as that of n and k, respectively with the variation of x values and heat treatment. This due to that the variations of  $\varepsilon_1$  mainly depends on the value of  $(n^2)$ , because of the smaller values of  $(k^2)$  comparison with  $(n^2)$ , while the imaginary part of the dielectric constant mainly depends on (k) values which were related to the variations of absorption coefficient.

# 4-Conclusions

- The X-ray diffraction observed that the prepared films were of poly crystalline kind of the hexagonal.
- 2. It is observed from the transmission spectrum that the thin films surfaces and thickness was uniform.
- 3. The refractive index in SnS<sub>2</sub> and SnS<sub>2</sub>Cu films varies with varying the composition by non-linear relation. The refractive index increases after heat treatment due to the increasing in the films compactness.

The variation of real and imaginary parts of dielectric constant have similar trends as for refractive index and extinction coefficient respectively according to Maxwell's equations

4.



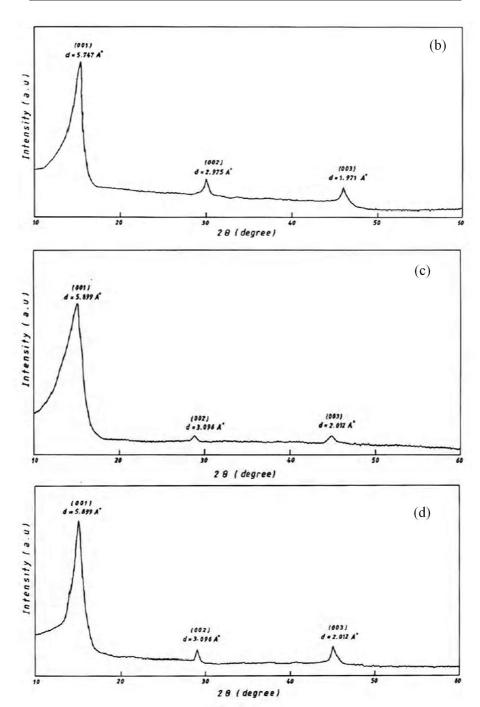


Fig. (1) XRD pattern for a-SnS2films as-deposited and, b-SnS2 films annealed to 473K ,c- SnS2:4%Cu films as-deposited ,d- SnS2:4%Cu films annealed to 473K.



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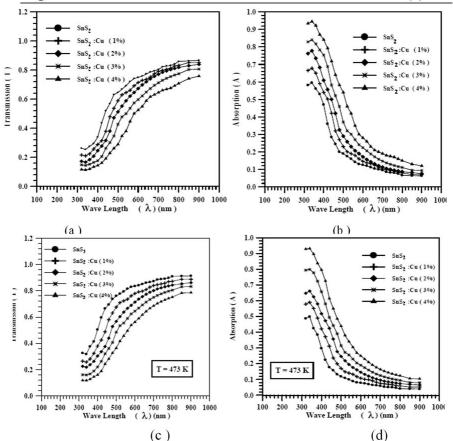


Fig.(2) Transmission spectrum of a-SnS $_2$ :0,1,2,3,4%Cu films as-deposited and,c-SnS $_2$ :0,1,2,3,4%Cu films as-deposited to 473K ,absorption spectrum of b- SnS $_2$ :0,1,2,3,4%Cu films as-deposited , and d- annealed to 473K.

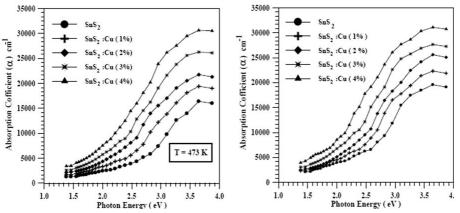


Fig. (3) The absorption coefficient verses wavelength for  $SnS_2:0,1,2,3,4\%$ Cu  $CdSe_{1-x}Te_x$  thin film at different composition (a) as-deposited (b) annealed.

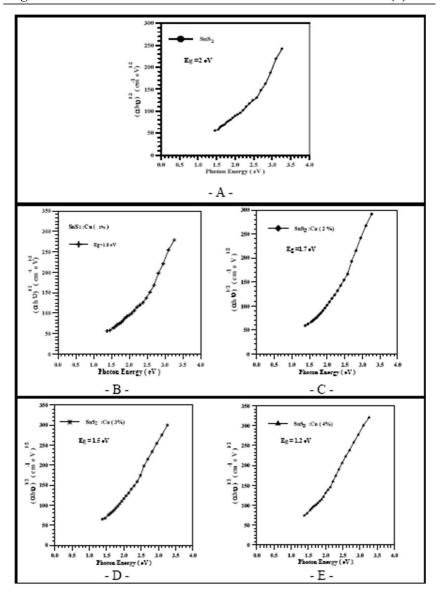


Fig. (4-a) The variation of  $(\alpha h v)^{1/2}$  with photon energy for  $SnS_2$ :Cu film at different composition (A) pure - (B) 1%Cu ,(C)- 2%Cu ,(D)-3% Cu(E)-4%Cu.

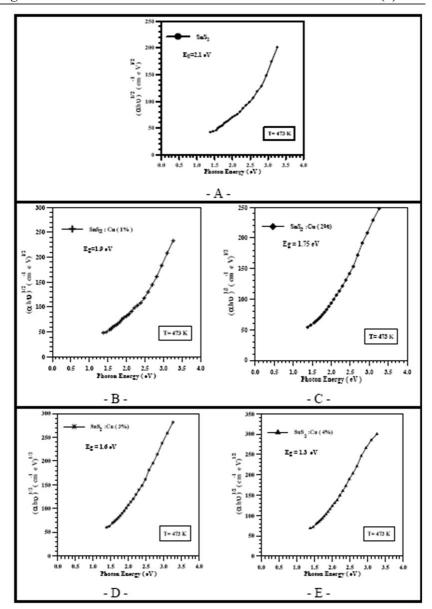


Fig. (4-b) The variation of  $(\alpha hv)^{1/2}$  with photon energy for SnS<sub>2</sub>:Cu film after annealing (A) pure - (B) 1%Cu ,(C)- 2%Cu ,(D)-3% Cu(E)-4%Cu.

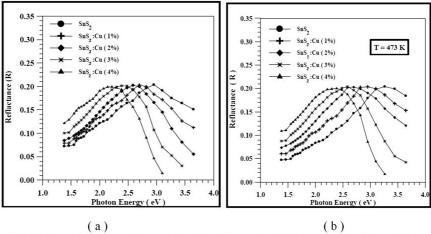


Fig.(5) The refractive index as a function of wavelength for as a-deposited and b-annealed  $SnS_2$ .Cu thin films.

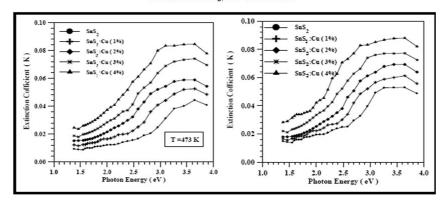


Fig.(6) The extinction coefficient versus wavelength for as-deposited and annealed

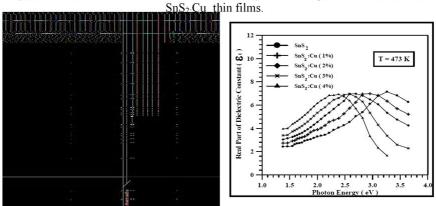
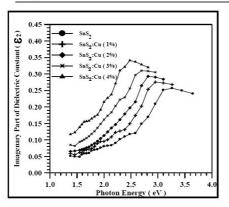


Fig.(7) The real part of dielectric constant as a function of wavelength for as-deposited and annealed  $SnS_2$ :Cu thin films



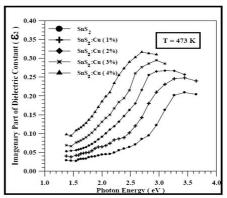


Fig.(8) The imaginary part of dielectric constant versus wavelength for as-deposited and annealed SnS<sub>2</sub>C thin films

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# الخواص SnS2:Cuالرقيقة المحضرة بطريقة التحلل الكيميائي الحراري التركيبية والبصرية لأغشية

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#### الخلاصة:

حضرت أغشية رقيقة من ثنائي كبريتيد القصدير ( SnS<sub>2</sub> ) النقية والمشوبة بالنحاس بنسب التشويب (1,2,3,4 )% بنقنية الرش الكيميائ (SnS2:Cu) بنسب التشويب (1,2,3,4)% بنقنية الرش الكيميائي الحراري (شواري) (603K) على قواعد من الزجاج مسخنة بدرجة حرارة (603K) وبسمك  $\mu m$  وبعد ذلك تمت معاملة الأغشية حرارياً تحت ضغط واطئ  $(0.7\pm0.02)$  بدرجة حرارة K فرادة المدة ساعة واحدة درس تأثير كل من التشويب بالنحاس والمعاملة الحرارية (التلدين بدرجة حرارة X 473 على بعض الخصائص الفيزيائية للأغشية المحضرة (التركيبية والبصرية أظهرت فحوصات الأشعة السينية بأن الأغشية المحضرة ذات تركيب متعد التبلورPolycrystalline)ومن النوع السداسي (Hexagonal) تبين من الدراسة البصرية التي شملت أطياف الإمتصاصية والنفانية في المدى الطيفي-300 (900) أن قيم معامل الإمتصاصα للأغشية النقية والمشوبة اكبر من cm 104-1)nm ) كما وجد أن الإنتقالات الْألكترونية عند حافة الإمتصاص الأساسية كانت من نوع الإنتقال غير المباشر بنوعية المسموح والممنوع، وإن قيمة فجوة الطاقة البصرية في حالة الإنتقال غير المباشر المسموح تقل من eV 2الى eV عند نسب التشويب eV على التوالي بينما نقل من %(1.2,1.5,1.7,1.8) الىeV في حالة (1.1.3.1.5.1.7) الانتقال غير المباشر الممنوع ولنسب التشويب نفسها، وكذلك وجد أن قيمة فجوة الطاقة للأغشية النقية والمشوبة تزداد بعد التلدين كما استعملت أطياف الإمتصاصية والنفاذية أيضًا في إيجاد الثوابت R ومعامل الانكسار n والانعكاسية  $\epsilon$  ومعامل الخمود  $\epsilon$  ومعامل الانكسار n والانعكاسية والجزء الخيالي لثابت العزل2ع ووجد أن جميعُ الثوابت البصرية تتأثر بتغير نسب ، التشويب، كما إنها تبدي تأثرًا بعد معاملة الأغشية حراريا.