

Polyphenol Profile, Antioxidant Potential, Vitamin C and Mineral Contents of Crude Parsley Leaves Extract (*Petroselinum crispum*) and Its Isolated Apigenin: A Comparative Study

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Abstract

Petroselinum crispum or parsley is a typical culinary vegetable that is used as a flavouring and aromatic food additive. The aim of this research was to purify a major flavonoid present in parsley leaves, apigenin and evaluate the polyphenol, mineral and vitamin C contents in parsley leaves extract (PLE); in addition, the study compared antioxidant activities of ethanolic PLE with purified apigenin (AP), applying different non-enzymatic strategies using ascorbic acid (AA) as a reference. A complete screening for bioactive compounds in the ethanolic PLE was performed by HPLC analysis. The research outcomes indicated that purification of apigenin was successfully achieved by column chromatography with a high yield and purity; a single peak in the HPLC chromatogram confirmed this. The analysis revealed that contents of total phenolic and flavonoid in the ethanolic PLE were 75.91 ± 3.33 mg. g⁻¹ TA Eq and 15.61 ± 0.18 mg. g⁻¹ respectively; vitamin C is found to be 4.01 mg. g⁻¹. Our results indicated that among the trace minerals in parsley leaves, calcium and sodium documented the highest values; 14.13 mg. g⁻¹ and 8.51 mg. g⁻¹, while copper and manganese had the lowest values; 0.02 mg. g⁻¹ and 0.012 mg. g⁻¹ respectively. Finally, our findings demonstrated that purified AP had superior potential than ethanolic PLE for reducing power, total antioxidant, nitric oxide and superoxide scavenging activities. In contrast, the ethanolic PLE exhibited greater antioxidant activity than AP towards suppressing the DPPH and hydroxyl radicals.

Keywords: Antioxidants, Apigenin, Chromatography, Parsley, Vitamin C.

Introduction

Medicinal plants with therapeutic and medicinal properties have a variety of pharmacological effects on human and animal health; therefore, it is regarded as essential for the existence of life^{1, 2}. The World Health Organization (WHO) has assessed roughly 80% of individuals in developing nations relies on natural medication to address their healthcare needs. Most of plants have effective functional ingredients against diseases but some plants are too rich with these chemicals and are favored for the treatment of ailment³. Nowadays, pharmacological plants are

acknowledged as an effective source of newly discovered compounds of therapeutic value. When compared to modern medicine, herbal medicines are still widely used, especially the former fail to be free from adverse effects and fail to cure some chronic conditions^{4, 5}. *Petroselinum crispum* (Mill.) or parsley, as a one of an important medicinal plant, belong to the family *Apiaceae* or *Umbelliferae*. It is produced in vegetable garden, which is frequently added to food for flavoring and aromatic additive⁶. Generally, parsley contains numerous natural bioactive

components. For instance, flavonoids and polyacetylenes are main components of parsley⁷. Furthermore, it contains a number of crucial nutrients, including vitamins "C, A, B, K, tocopherols" and ergosterol. Minerals (iron, zinc, magnesium, calcium, sodium, potassium, phosphorus), furanocoumarins, carotenoids and fatty acids (linolenic and palmitic acid), are abundant in parsley⁷⁻⁹. Interestingly, *P. crispum* is being suggested for various therapeutic purposes in everyday medicine¹⁰. In this context, the effective diuretic and antispasmodic activities of parsley was investigated. Besides, the bioactive constituents in parsley are also associated with the variety of its beneficial effects, such as hepatoprotective, anti-diabetic, and analgesic characteristics. Remarkably, within numerous medical contexts, parsley leaves have demonstrated their therapeutic efficacy as medicinal herbs with anti-inflammatory, anti-anemic, anti-hyperlipidemic, anti-cancer, anti-hypertensive, and anticoagulant effects, which is mainly related to their component diversity⁸. Moreover, parsley leaves have been linked to a variety of formulations and bioactive compounds, including those for the treatment of immunological disorders, diabetes, antibacterial activities, as well as in the treatment of stroke, chronic bronchitis, dyspepsia, allergies^{7, 11, 12}. Along with the beneficial roles of parsley, the kaempferol, quercetin, and other flavonols, as well as luteolin and apigenin (glycosylated flavones) are the two main flavonoids

type compounds present in parsley. These classes of compound possess antioxidant properties and high ability for neutralization of free radicals. Thereby, exerts its sanogenous effects in the prevention of ailments linked with oxidative stress^{13, 14}.

Apigenin is a kind of secondary metabolites that is considered as one of main flavonoids found in large quantities in a set of natural provenances, for instance fruits and vegetables. Apigenin is identified as 4', 5, 7-trihydroxy flavone, possessing a chemical composition of C₁₅H₁₀O₅¹⁵. Apigenin is particularly abundant in celery, parsley, chamomile, vine spinach, oregano and artichokes. The richest source of apigenin among these is dried parsley¹⁶. Interestingly, numerous biological effects including their anti-inflammatory, antioxidant, anticancer, antigenotoxic, anti-allergic, neuroprotective, cardioprotective and antimicrobial have been linked to flavones and its synthetic derivative, apigenin¹⁷. The present study basically aimed to: (i) purify the major flavonoid, apigenin from parsley leaves using column and thin-layer chromatography techniques; (ii) achieve a complete screening (qualitatively and quantitatively) for the main ingredients found in parsley leaves using HPLC chromatography; (iii) compare the antioxidant activity of crude parsley extract with its purified compound (apigenin) by applying different non-enzymatic methods; (iv) profile polyphenol (total phenolics and flavonoids), vitamin C and mineral contents in ethanolic PLE.

Materials and Methods

Plant Collection

Fresh parsley leaves were gathered from a local market in Dohuk city, Kurdistan region, Iraq. Parsley was first cleaned using tap water and followed by running demineralized water and then brought into the air in a dark room to dry. The dried parsley leaves were then crushed by a mill and stored in dark color containers at -20 °C until the day of use.

Methods

Preparation of Parsley Leaves Extracts (Crude extract)

About 20 g dried powder of parsley leaves were placed in a conical flask wrapped with aluminum foil containing 250 mL of 70% ethanol and soaked overnight, followed by stirring at room temperature for 24 hrs. using magnetic stirrer, after that the extract was filtrated by Buchner under vacuum utilizing filter paper Whatman No.1. After that, a rotary evaporator is utilized to concentrate purify

solution at 45°C until it was completely dry. The parsley powder kept in a dark container and frozen until using for subsequent experiments.

Qualitative and Quantitative Analysis of Bioactive Compounds in Parsley Leaves Using HPLC Chromatography

In the next set of experiments, the HPLC analysis was carried out for quantification of individual phenolic compounds (crude extract and purified AP), in order to determine the bioactive constituents and purity of isolated AP respectively. The study was carried out using reversed phase HPLC and a Zorbax Eclipse Plus-C18-OSD (25cm/4.6mm) column with a SYKAMN HPLC chromatographic system outfitted with a UV detector. The column was heated at 30°C in temperature and injected volume (100 µL) of samples and standards automatically took place employing autosampler. The spectrum was monitored at 280 nm¹⁸. The identity of each peak in

the HPLC chromatography's of injected PLE and purified AP were compared with the peaks of authentic standard. Furthermore, the amount of each peak was also determined using the following formula:

$$\text{Conc. (ppm)} = \frac{\text{Area (sample)} \times \text{standard conc.}}{\text{Area (Standard)}} \times \frac{\text{NO. of dilution}}{\text{sample weight}}$$

Isolating Flavonoids from Parsley Leaves

The isolation of flavonoids from parsley leaves was conducted utilizing the methodology of¹⁹, with certain changes. In this method, green thick solution of total flavonoids was obtained and used for subsequent apigenin separation.

Separation of Apigenin from Flavonoids by Chromatography Techniques

In order to separate apigenin from total flavonoids, two chromatography techniques were used. Column chromatography was applied firstly to isolate the bioactive compound, apigenin (AP) from flavonoid²⁰. In this particular approach, a column measuring 60x2 cm was utilized, which was packed with slurry consisting of silica gel-H with a mesh size ranging from 60-120 μ (Hi media, Mumbai). After that, the fractions were eluted using a gradient solvent system consisting; n-butanol, water, and acetic acid (12: 2: 1 v/v/v). For detection AP in elutes, thin-layer chromatography technique (TLC) was performed for each fraction as primary detection of apigenin and compared the retention factor (R_f) of each spot with R_f value of standard apigenin²¹.

Analysis of Polyphenol Profile and Vitamin C in Parsley Leaves Extract

Each test in this section was performed in triplicates as well as for standard solutions.

Estimation of Entire Phenolic Constituents

A total amount of phenolic constituents in PLE were measured via Folin-Ciocalteu method. The total phenolic concentration in this method was estimated depending on the tannic acid standard curve²².

Estimation of Entire Flavonoid Constituents

The modified aluminum chloride method introduced by²³, was applied for quantification of total flavonoids in PLE. Different concentrations of quercetin were used also for preparation of quercetin

standard curve, which was used to determine quercetin equivalents in sample.

Determination of Ascorbic Acid (Vitamin C) Contents

Ascorbic acid concentration in PLE assessed spectrophotometrically utilizing a method conducted by²⁴ with certain changes. In this approach, a standard curve was plotted using reference compound; AA to determine its concentration in the extract. The calculation the quantity of AA equivalents for each gram of extract (mg.AAE.g⁻¹) was performed utilizing the formulation that reported by²⁵.

$$V = C \times V/M$$

V= AA content (mg.AAE.g⁻¹), C= Ascorbic acid Concentration extracted from standard curve (mg.ml⁻¹), V= Volume of PLE (ml) and M= PLE weight (g)

Trace Minerals Analysis

The concentrations of trace minerals (Ca, Na, Mg, Cu, Mn and Fe) in parsley leaves were determined using a wet acid digestion method and the analysis was carried out by Atomic Absorption Spectroscopy (AAS)²⁶.

Antioxidant Activity Assessments of Parsley Leaves Extract and Purified Apigenin

The antioxidant activity of PLE and its purified AP *in vitro* was carried out using different methods. A stock solution (1mg/ml) of each sample and standard was prepared and dissolved in distilled water (D.W.) and subsequently used for antioxidant analysis. Each test was done on three times as well as for standard solutions.

Reducing Power Assay

A reducing power of PLE and its purified AP has been estimated by the procedure described by²⁷, and ascorbic acid functioned as a basis for comparison in this test.

Comprehensive Antioxidant Activity Assay

In this test, a spectroscopic phosphomolybdenum procedure was done for determination of comprehensive antioxidant capacities of PLE and purified AP as described by²⁸. Ascorbic acid functioned also as a basis for comparison in this assay.

Nitric Oxide Suppressing Assay

In order to evaluate the suppression of nitric oxide by PLE and purified AP, the approach introduced by²⁹

was used. In this assay, vitamin C was utilized as standard antioxidant for comparison. Further, control samples were also prepared by the same procedure without plant samples. The proportion (%) of nitric oxide radicals that are inhibited was established using the formula: % Inhibition of NO radical = $[(A_o - A_1) / A_o] \times 100$

The absorptions of control and sample are referred as A_o and A_1 , respectively.

Hydroxyl Radical Scavenging Assay

The capability of PLE and purified AP to inhibit hydroxyl radicals was verified by the procedure done by³⁰. The same context as previously mentioned for nitric oxide was applied for calculation of the % of inhibition.

Ferrous Ion Chelating Assay

The capacity of PLE and its bioactive component, AP to sequester the ferrous ions (Fe^{+2}) was estimated by the method of³¹, which is outlined by³². Vitamin C standard solution was used in this assay for comparison. Ferrous ion chelation capacity was calculated as follow: Bound Fe^{2+} (%) = $[(Ac - As) / Ac] \times 100$

Superoxide Anion Suppressing Analysis

Results and Discussion

Qualitative and Quantitative Analysis of PLE by HPLC

HPLC technique was applied in the present study for qualitative investigation of crude PLE as shown in HPLC chromatogram Fig. 1. The polyphenolic compounds found in crude alcoholic PLE berries compared with retention time of the standard phenolic compounds. The analysis revealed that the major bioactive components found in parsley leaves are apigenin, luteolin, p-Coumaric acid, epicatechin, kaempferol and quercetin. The main compounds and their corresponding concentration are presented in Table 1. As documented in Table 1, the highest polyphenol compound in PLE was epicatechin (45.99 ppm) followed by apigenin (41.56 ppm), while the other bioactive compounds were in lower concentrations. Due to the presence of a wide spectrum of active substances as confirmed by HPLC analysis, parsley therefore may be used as a medicinal plant with a diversity of antihypertensive and anticoagulant anti-inflammatory, anti-anemic, anti-hyperlipidemic and anticancer activities^{8, 37}.

In this experiment, the method of³³ was used for estimation the potential of PLE as well as AP in suppressing of superoxide anion radicals. In a similar fashion to the calculation of nitric oxide, the percentage of superoxide anion suppressing activity is determined.

Activity to scavenge superoxide anion radicals (%) = $[(A_c - A_s) / A_c] \times 100$

Absorptions of control and sample are denoted by A_c and A_s , respectively.

DPPH (1, 1-diphenyl-2-picrylhydrazyl) Suppressing Assay

The activity of the specimens to suppress DPPH radicals was assessed as stated by³⁴. In this assay, the blank and control solutions were also prepared and the percentage of suppressing activity (AA%) was calculated in accordance with³⁵.

AA% = $100 - [(O.D_{\text{sample}} - O.D_{\text{blank}}) / O.D_{\text{control}}] \times 100$
O.D = Optical density

Data Analysis

Results from each experiment were given as mean \pm SE (standard error) values, which were carried out in triplicate. Duncan's multiple range analysis was used to separate the means from one another and an analysis of variance was used to assess the results ($P < 0.05$). SPSS was used to analyze the outcomes³⁶.

Furthermore, the high number of flavonoids with antioxidant power and its capacity to suppress free radicals possibly the cause of parsley's sanogenous effects^{13, 38}.

Table 1. HPLC results for the qualitative and quantitative analysis (ppm) of crude PLE compared to standard phenolic and flavonoid compounds.

Standard compound (Rt/min.)	Alcoholic PLE (Rt/min.)	Concentration (ppm)
Apigenin, (7.22)	7.12	41.56
Luteolin, (7.72)	7.79	20.66
p-Coumaric acid, (8.02)	8.26	20.58
Epicatechin, (8.95)	8.89	45.99
Kaempferol, (9.09)	9.00	33.56
Quercetin, (13.92)	13.89	31.78

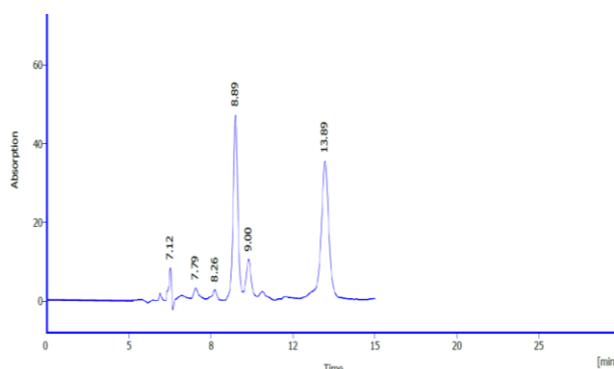


Figure 1. HPLC Chromatograms for main isolated bioactive compounds and their retention times accordingly (apigenin:7.12 min., Luteolin:7.79 min., p-Coumaric acid :8.26min., epicatechin:8.89min., kaempferol:9.00min and quercetin:13.89min).

Isolation of Flavonoids from Parsley Leaves

Due to its practicality, affordability, and availability in different stationary phases, both column chromatography and thin-layer chromatography (TLC) are still widely used³⁹. However, the parsley leaf extract was used in the current work to isolate apigenin as mentioned in section 2.4 and 2.5. The results showed that the isolation of apigenin was successfully achieved by column chromatography with a high yield (1.5g) and purity. Furthermore, thin-layer chromatography was applied individually on all fractions obtained from the column with the same solvent system to evaluate the R_f value on TLC plate and compare it with standard apigenin. The R_f for the isolated AP was 0.86, which is close to the R_f of standard AP (0.84). As well, the HPLC analysis Figs. 2A and B for purified AP showed the characteristic single peak in HPLC chromatogram at 4.27 min. retention times (R_t) compared with standard apigenin compound (4.12 min). These data provided further verification of the purity of isolated apigenin using chromatographic strategies.

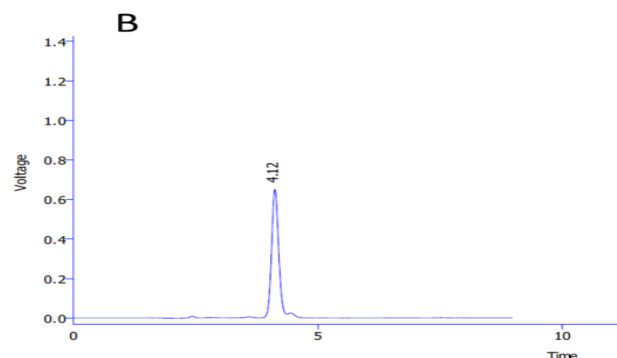
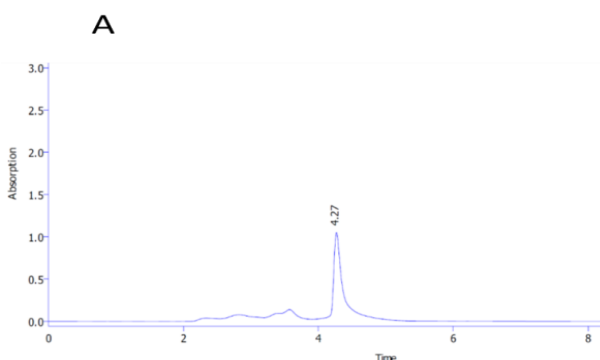


Figure 2. HPLC Chromatogram for A) Isolated apigenin from crude parsley leaves, B) HPLC chromatogram for the authentic apigenin injection.

Polyphenol Profile and Vitamin C Contents in Parsley Leaves Extract

Total Phenol and Flavonoids Contents

In view of the current study, the quantity of total phenolic components in ethanolic PLE was 75.91 ± 3.33 mg/g (TA Eq). In contrast, our data showed that the amount of total flavonoid in PLE was 15.61 ± 0.18 mg/g. Comparing our findings to a previous study of⁴⁰, the total phenolic contents were lower (21.63 ± 1.04 mg/g), while the total flavonoid was comparable to our result, with the study of¹¹ demonstrated that both contents were higher, possibly due to usage of different extraction conditions. Interestingly, the valuable amount of total phenolic and flavonoid constituents in parsley leaves may reflect their radical scavenging activity as well as potential chemo-preventive/ protective roles in maintaining human health and illness prevention⁴¹. Thereby, it possess a variety of biological activities, including organ protection, hypoglycemic, lipid-lowering and anti-inflammatory effects^{42, 43}.

Ascorbic Acid Contents

Vitamin C contents in ethanolic PLE were 4.01 ± 0.33 mg/g. According to several studies^{44, 45}, vitamin C is the most significant biological component of parsley leaves and one of excellent supply of this vitamin of the daily requirements^{46, 47}. In this context, vitamin C is very essential for modulating the immune system and able to protect vital biomolecules from destruction resulting from oxidants, toxins, and pollutants by donating electrons because it is a potent antioxidant⁴⁸. Furthermore, vitamin C also involved in the creation of amidated peptides including vasopressin and catecholamine hormones⁴⁹ and

collagen synthesis⁵⁰. Strikingly, vitamin C contents in parsley leaves may contribute to all of aforementioned health benefits for human.

Trace Minerals

The data shown in Table 2 indicate to the mineral analysis of parsley leaves. The analysis showed that among all minerals detected in PLE, calcium was highest (14.13 mg/g), followed by sodium (8.51 mg/g). A study by⁵¹ revealed that, the nutritional value of parsley was $1986 \pm 40 \mu\text{g} \cdot \text{g}^{-1}$ of sodium. As shown in Table 2, magnesium was present in a reasonable amount ($2.14 \pm 0.01 \text{ mg/g}$). In contrast to⁵², which indicated that Mg in parsley leaves was below the detection limit, possibly due to unsuccessful or un-sufficient digestion strategy. In our study, copper (Cu) and manganese (Mn) had low concentrations in parsley leaves, about 0.012 and 0.02 mg/g, respectively. Finally, the iron content of the parsley leaves under examination was 0.22 mg/gm. Extensive evidence exists about the usefulness of minerals for handling of common health problems. In this context, the mineral levels in blood and cellular fluids are of great importance for bones health and cardiac muscle^{53, 54}, nutrition⁵⁵, as enzyme cofactors⁵⁶, regulation of neuronal activity, immunological response, energy generation, and antioxidant defense^{57, 58}, and development of connective tissues²⁶. Therefore, parsley may serve as source of trace minerals to support for normal bodily functions.

Table 2. Trace Elements Concentrations of PLE (mg/g).

Elements	Amount (mg/g) dry weight
Calcium	14.13 ± 0.05
Sodium	8.51 ± 0.01
Magnesium	2.14 ± 0.01
Copper	0.02 ± 0.01
Manganese	0.012 ± 0.01
Iron	0.22 ± 0.09

Antioxidant Activity of Parsley Leaves and Purified Apigenin

The parameters measuring antioxidant activity were presented as mean \pm S.E of three replicates.

Reducing Power of PLE and Purified AP

In this study, the ability of PLE and AP to convert ferric ion to ferrous ion used as a measure for reductive power of both compounds compared to reference compound (AA), which is a well-known

powerful reducer. As displayed in Table 3, both compounds showed the powerful of reducing properties. This finding may suggest that the reducing power of parsley leaves might be due to presence of the biologically active substances such as antioxidant polyphenols, minerals, essential oils, and pigments.

Our data showed also that with raising the concentrations of PLE and purified AP, the reducing power increased accordingly. This observation may be because of increasing the antioxidants' capacity to donate electrons. Furthermore, apigenin seems to have stronger reducing activity than PLE, and this difference was statistically significant. The reason can be attributed to the chemical nature of apigenin, which is trihydroxy phenols that have a greater number of hydroxyl groups are joined to ring, these group act as more powerful reducing agent and considered as good proton donors⁵⁹. Research published by¹³ showed that parsley roots have reducing activity as well.

Table 3. Dose-dependent Reducing Power of PLE and Purified AP.

concentrations (ppm)	PLE	AP	AA
23.53	$0.02^a \pm 0.01$	$0.07^{ab} \pm 0.01$	$1.85^c \pm 0.01$
70.59	$0.07^{ab} \pm 0.01$	$0.09^{ab} \pm 0.01$	$2.84^d \pm 0.01$
117.64	$0.13^c \pm 0.01$	$0.18^b \pm 0.01$	$2.94^d \pm 0.1$

The numbers in the identical column and raw accompanied by similar alphabetical characters, mean no significant changes ($P > 0.05$).

Comprehensive Antioxidant Activity of PLE and Purified AP

Table 4 displays the total antioxidant activity of the PLE, AP in comparison to AA. The overall antioxidant capacity of PLE and AP are likely due to the flavonoids, which are widely distributed in parsley leaves. These substances may primarily act as antioxidants, stabilizing free radicals by resonance or through oxidation-reduction processes as reducing agents⁶⁰. Hence, it provides a potent intracellular defense against "reactive oxygen species (ROS) and oxidative stress" related disease.

The findings revealed that purified AP has higher overall antioxidant properties than PLE. This may be explained by the high potential of AP for being a proton donor and hence potent antioxidant compound compared with PLE⁶¹. Our results

concerning the overall antioxidant capacity of PLE and AP were dose-dependent. Compared with vitamin C, significant changes were noticed ($p < 0.05$) in total antioxidant capacity when both, PLE and AP compared with AA, which was also dose dependent. Similar trend of total antioxidant activity of PLE was noticed from study done by¹¹, using different solvent system.

Table 4. Comprehensive Antioxidant Activity of PLE and Purified AP.

concentrations (ppm)	PLE	AP	AA
30.3	0.44 ^a ± 0.01	0.48 ^{ab} ± 0.04	1.84 ^e ± 0.07
60.6	0.54 ^b ± 0.04	0.63 ^c ± 0.02	2.2 ^f ± 0.08
90.9	0.61 ^c ± 0.01	0.72 ^d ± 0.01	2.54 ^g ± 0.07

The numbers in the identical column and row accompanied by similar alphabetical characters, mean no significant changes ($P > 0.05$).

Scavenging of Nitric Oxide (NO) by PLE and Purified AP

Based on the data given in Table 5, the *in vitro* NO scavenging assay suggests that both, PLE and purified AP have the capacity to scavenge NO radicals as displayed by high percentage (%) of inhibition at various concentrations compared to vitamin C. This scavenging activity against NO could be explained by the fact that the secondary metabolites in parsley are responsible for the antioxidant properties. As well, the results illustrated in Table 5 showed that AP exhibited a higher and significant degree of nitric oxide scavenging activity ($p < 0.05$) in comparison to PLE. Indeed, the impact of phytochemicals on NO suppressing may be of great interest for their therapeutic applications. In this regards, multiple evidence showed that an activation of pro-inflammatory mediators linked to both acute and chronic inflammation could result in tissue damage and NO overproduction⁶². As a result, NO inhibitory medications may be useful for treating the inflammatory response⁶³.

Table 5. Nitric Oxide Suppressing Activity of PLE and Purified AP (% Inhibition).

concentrations (ppm)	PLE	AP	AA
66.67	50.22 ^{abc} ± 10.04	64.15 ^{de} ± 4.3	59.46 ^{bcd} ± 6.72
200	43.59 ^a ± 8.4	49.93 ^{ab} ± 3.8	52.73 ^{abcd} ± 4.55
333.33	62.94 ^{de} ± 7.35	63.86 ^{de} ± 10.55	81.2 ^f ± 7.25

The numbers in the identical column and row accompanied by similar alphabetical characters, mean no significant changes ($P > 0.05$).

Hydroxyl Radical Scavenging Activity of PLE and Purified AP

Table 6 demonstrates the impact of PLE and purified AP on the suppressing of OH radicals. The results of present investigation have substantiated the inhibitory influence of PLE and purified AP on generation of OH radicals that is relied on the dosage administered. This observation maybe related to secondary metabolites presence in PLE especially kaempferol, which is known to possess a higher ability to scavenge OH radicals⁶⁴. On the other hand, the inhibitory effect of AP is likely due to the presence of high degree of hydroxylation on their aromatic rings, which may suggest its higher contribution in the inhibition of OH radicals' formation⁶¹. Moreover, the results in Table 6 showed also that PLE was more effective than AP in scavenging of OH radicals, and both possess lower scavenging activity compared to ascorbic acid ($p < 0.05$). Similar findings have been established regarding the ability of numerous plant extracts and flavonoids, such as naringin and mangiferin in hydroxyl free radicals' neutralization^{65, 66}. Along with the monitoring of the obtained data, plant extracts can interact with OH radical to reduce their harmful effects to cell and its components. Therefore, the assessment of OH radical suppressing activity of vegetable matrices holds significant importance from biological perspective, as it aids in the identification of efficacious extracts that are suitable for pharmacological purposes⁶⁷.

Table 6. Hydroxyl Radicals Scavenging Activity of PLE and Purified AP (% Inhibition).

concentrations (ppm)	PLE	AP	AA
16.66	28.69 ^{ab} ± 2.01	21.56 ^a ± 2.74	27.2 ^{ab} ± 2.5
25	52.34 ^{def} ± 4.47	36.98 ^c ± 3.73	54.68 ^{ef} ± 7.9
33.33	69.29 ^{hi} ± 3.14	63.03 ^{gh} ± 3.41	82.42 ^j ± 1.52

The numbers in the identical column and raw accompanied by similar alphabetical characters, mean no significant changes ($P > 0.05$).

Ferrous Ion Chelating Ability of PLE and Purified AP

Fe^{+2} binding ability (%) of the assays reviewed in Table 7. According to our findings, both, PLE and AP have iron chelation effect and so, preventing the reaction between the reagent and the ferrous complex. This effect may be consistent to the excessive antioxidant contents of parsley leaves. Where these antioxidants functioned as sequestering tools by decreasing the potential of oxidation-reduction reactions and stabilizing the metal ion's oxidized state^{68, 69}. Besides, the studies introduced by⁶¹ documented the efficacy of apigenin as a robust antioxidant and chelator of redox active metals both in vitro and in vivo. The data obtained demonstrated that activity of metal chelator of PLE and AP exhibited a gradual increase with increasing concentrations, thereby demonstrating the capacity of the plant extract components to form complexes with ferrous ions. Furthermore, our finding indicates that the both PLE and AP have a stronger tendency for chelating iron than ascorbic acid ($p < 0.05$), while, no statistic significant were noted in this effect between PLE and AP. Our results are compared to those of⁷⁰ who used one concentration and different solvent system and¹³ who used root juice of parsley in contrast to our study using 70% ethanol.

Table 7. Fe^{+2} Chelating Ability (%) of PLE and Purified AP.

concentrations (ppm)	PLE	AP	AA
17.42	37.8 ^{bc} ± 0.88	37.88 ^{bc} ± 1.76	26.67 ^a ± 4.58
34.84	46.15 ^{cd} ± 1.01	44.69 ^{cd} ± 2.83	44.61 ^{cd} ± 6.59
52.3	65.27 ^e ± 1.54	69.23 ^e ± 2.81	63.96 ^e ± 9.5

The numbers in the identical column and raw accompanied by similar alphabetical characters, mean no significant changes ($P > 0.05$).

Scavenging of Superoxide Anion by PLE and Purified AP

Table 8 summarizes the percentage of inhibition of superoxide radical by PLE and purified AP comparing to ascorbic acid as standards. The results clearly indicated the contribution of both, PLE and AP in the scavenging of superoxide radicals in a dose-dependent way. While, the purified AP has stronger scavenging effect than PLE with a significant difference ($p < 0.05$) and both have a weaker inhibition than ascorbic acid. The reason behind this scavenging effect might be due to the parsley contents of kaempferol, quercetin as well as glycosylated flavones type compounds, which all possesses antioxidant properties or to the ability of AP to bind with superoxide anion¹³. Nevertheless, research done by⁷¹ investigated the effect of parsley leaves for enhancement of antioxidant enzymes activities; specifically, erythrocyte glutathione reductase (GR) as well as superoxide dismutase (SOD) and discovered that apigenin is mostly accountable for this effect. At pathological level, the super oxide anion considered harmful, as they can induce apoptosis, necrosis, and autophagic cell death⁷². Thus, our data concerning superoxide scavenging by PLE and AP may shed a light on the possible therapeutic uses of these compounds against several pathological events.

Table 8. Superoxide Anion Radical Scavenging Activity of PLE and Purified AP (% Inhibition).

concentrations (ppm)	PLE	AP	AA
50	33.02 ^a ± 1.1	44.77 ^c ± 2.7	52.7 ^d ± 2.82
100	38.85 ^b ± 2.83	51 ^d ± 0.67	67.7 ^e ± 4.62
200	49.72 ^{cd} ± 0.93	65.21 ^e ± 1.01	82.39 ^f ± 3.26

The numbers in the identical column and raw accompanied by similar alphabetical characters, mean no significant changes ($P > 0.05$).

Suppressing of DPPH Radicals by PLE and Purified AP

The DPPH free radical capturing assay is an effective technique used in this research to assess scavenging potential of both PLE and purified AP compared to ascorbic acid. In the current work, the reduction in

the optical density was induced by both compounds, and thus, serves as a measure of the effectiveness of both specimens in the removal of harmful radicals. Our data showed also that increasing the concentration of PLE and AP in the reaction mixture accompanied with the elevation of the scavenging activity (dose dependent) Table 9. This effect might be ascribed to the phytochemical nature of specimens, which facilitated their ability to perform as hydrogen donors and quenchers of singlet oxygen. The mechanism of apigenin action includes: chelating of metals; scavenging of free radicals and amplification enzymatic and non- enzymatic antioxidants ⁶¹. Data established in Table 9 indicated that PLE has stronger and significant extent to reduce DPPH when compared with purified AP ($P < 0.05$). Apparently, this might be associated with the need to a variety of antioxidant species for scavenging of DPPH radicals rather than AP alone, compared with

ascorbic acid, PLE and AP showed lessened abilities for changing the stable DPPH radicals to DPPH-H form. Our results compared to studies of ^{73, 74} which also discovered the same finding but in methanolic and aqueous extract of parsley leaves.

Table 9. The Percentage of DPPH Radicals Inhibition (%) by PLE and Purified AP.

concentrations (ppm)	PLE	AP	AA
36.49	5.18 ^b ± 0.55	1.52 ^a ± 0.7	97.56 ^g ± 0.7
72.99	14.61 ^d ± 0.46	11.11 ^c ± 2.3	98.33 ^g ± 0.35
145.98	40.94 ^f ± 1.62	22.91 ^e ± 1.73	98.71 ^g ± 0.86

The numbers in the identical column and row accompanied by similar alphabetical characters, mean no significant changes ($P > 0.05$).

Conclusion

The findings of this research concluded that parsley leaves are considered as an excellent source of antioxidants (phenolic, flavonoid, and vitamin C contents). Regarding to trace elements analysis of parsley leaves, both calcium and sodium gave the highest concentrations followed by magnesium and iron while copper and manganese exhibited the lowest concentrations. Furthermore, this study demonstrated that apigenin (a major flavonoid) from parsley leaves is effectively purified using column chromatography, as evidenced by a single peak product in an HPLC chromatogram with a high yield and purity. By using non-enzymatic methods,

purified AP had more scavenging activity for reducing power, total antioxidant, nitric oxide, and superoxide scavenging than ethanolic parsley leaf extract. In contrast, the ethanolic PLE displayed stronger antioxidant activity than AP towards reducing the DPPH and hydroxyl radicals. Therefore, there is need to confirm these findings by doing *in vivo* animal model research to evaluate the role of this plant and its purified compound, apigenin in the scavenging of free radicals related diseases. As a result, it is necessary to better understanding and to screen medicinal plants for bioactive compounds as a basis for future pharmacological research.

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Authors' Declaration

- Conflicts of Interest: None.
- We hereby confirm that all the Figures and Tables in the manuscript are ours. Furthermore, any Figures and images, that are not ours, have been included with the necessary permission for re-publication, which is attached to the manuscript.
- No animal studies are present in the manuscript.
- No human studies are present in the manuscript.
- Ethical Clearance: The project was approved by the local ethical committee at University of Zakho.

Authors' Contribution Statement

A. A. M. and L. Y. M. contributed to the concept, design, and implementation of the research, and to the revision and proofreading of the manuscript.

A.A. I. performed the experiments, acquisition of data, analysis and interpretation of the research, to write the draft of the manuscript.

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تقييم محتوى البوليفينولات، فيتامين ج، المحتويات المعدنية بالإضافة الى الفعالية المضادة للأكسدة لمستخلص أوراق البقدونس الخام (*Petroselinum crispum*) بالمقارنة مع الأبيجينين المعزول منه

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الخلاصة

البقدونس أو *Petroselinum crispum* هو خضار طهي تقليدي يستخدم بشكل كبير كمكمل غذائي، كمنكهات او مضافات معطرة للأطعمة. تركزت اهداف الدراسة الى تنقية الفلافونيد الرئيسي الموجود في أوراق البقدونس، (الأبيجينين) وتقييم محتوى البوليفينول، فيتامين ج والعناصر المعدنية في مستخلص أوراق البقدونس بالإضافة الى مقارنة الأنشطة المضادة للأكسدة لمستخلص أوراق البقدونس الأيثانولي مع مركب الأبيجينين المنقى باستخدام طرق تحليلية مختلفة. علاوة على ذلك، اجراء تحليل كامل للمركبات الحيوية الأكثر وجودا في أوراق البقدونس باستخدام تحليل كروماتوغرافيا السائل العالي الأداء (HPLC).

أظهرت نتائج الدراسة الى امكانية تنقية مركب الأبيجينين من أوراق البقدونس باستخدام كروماتوغرافيا العمود بكفاءة ونقاوة عالية وذلك من خلال الحصول على حزمة واحدة نقية في كروماتوغرام ال HPLC. اشارت نتائج التحليل أيضا أن محتوى الفينول الكلي ومحتوى الفلافونيدات في مستخلص أوراق البقدونس الكحولي كان بحدود 75.91 ± 3.33 ملغ/غم، 15.61 ± 0.181 ملغم/غم على التوالي. بلغ محتوى فيتامين ج 4.01 ملغم/غم.

كما دلت نتائج تحليل المعادن النادرة، أن مستوى الكالسيوم والصوديوم كانتا الأعلى من بين المعادن المقاسة في أوراق البقدونس، بقيم بلغت 14.13 ملغم/غم و 8.51 ملغم/غم على التوالي بينما كان مستوى كل من عنصري النحاس و المنغنيز الاقل قيمة (0.02 ملغم/غم) و (0.012 ملغم/غم) على التوالي. وأخيرا، وجد من خلال النتائج أن للأبيجينين المنقى خواص مضاد اكسدة أعلى من المستخلص الكحولي للبقدونس من ناحية القوة الاختزالية، الفعالية المضادة للأكسدة الكلية، والقابلية على اكتساح جذور أكسيد النتريك و السوبرأوكسيد الفعالة. وعلى العكس من ذلك، كان للمستخلص الكحولي للبقدونس نشاط مضاد أكسدة اعلى من الأبيجينين المنقى في تثبيط جذور DPPH وجذور الهيدروكسيل.

الكلمات المفتاحية: مضادات الأكسدة، أبيجينين، كروماتوغرافيا، البقدونس، فيتامين ج.